

Tandem Reactions of 1,2,4-Oxadiazoles with Allylamines

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ABSTRACT



A reaction of 3-chloro-1,2,4-oxadiazoles with allylamine and diallylamine has been investigated. 3,3a,4,5-Tetrahydroisoxazolo[3,4-d]pyrimidines are produced through a tandem ANRORC/[3 + 2]cycloaddition pathway consisting of the addition of allylamine to the 1,2,4-oxadiazole, followed by ring opening, nitrene formation, and finally cycloaddition. 3-N-Allylamino-1,2,4-oxadiazoles were also obtained as minor products through a classical SNAr. Conversely, a reaction with diallylamine produces 3-N,N-diallylamino-1,2,4-oxadiazole and imidazoline through tandem SNAr/aziridination and nucleophilic ring opening.

ANRORC (Addition of a Nucleophile, Ring Opening, and Ring Closure) reactions represent a useful strategy for ring transformation of heterocyclic systems.¹ ANRORC rearrangements of six-membered heterocycles have been extensively studied by Van der Plas and co-workers,¹ while ANRORC reactions of five-membered systems concern electron-poor heterocycles such as 1,3,4-oxadiazoles,² 1,3,4-thiadiazoles,³ nitroimidazoles,^{2,4} bis(1,3,4-thiadiazol-2-yl)-1,3,5-triazinium halides,⁵ isothiazole,⁶ and isoxazoles.⁷

Taking advantage of 1,2,4-oxadiazole reactivity and their high tendency to rearrange,⁸ we have reported the

preparation of several fluorinated heterocycles (i.e., 1,2,4-triazoles,⁹ 1,2,4-oxadiazoles,¹⁰ 1,2,4-triazines,^{9b,11} 1,2,4-oxadiazinones,¹² indazoles¹³) via ANRORC reactions.

In this context, the presence of a strongly electron-withdrawing fluorinated group, linked at the C(5) of 1,2,4-oxadiazoles, has been so far considered as “*conditio sine qua non*” for the ANRORC reactivity. Only recently, we have demonstrated the ANRORC reactivity of 3-chloro-1,2,4-oxadiazoles also in the absence of C(5)-linked electron-withdrawing groups.¹⁴ In the latter case, it was suggested that the presence of a chlorine linked at C(3) of the

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oxadiazole ring leads to the formation of a chloro-oxime intermediate after the ring opening induced by the attack of the nucleophile. Considering that such an intermediate could represent a nitrile oxide precursor, we envisaged the possibility of a tandem ANRORC/[3 + 2]cycloaddition reaction with nucleophiles containing an olefinic moiety (Figure 1).

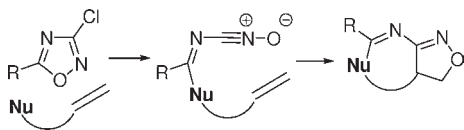


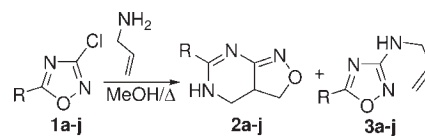
Figure 1. Schematic representation for the tandem ANRORC/[3 + 2]cycloaddition reaction.

Such an approach represents a new entry for the obtainment of heteroannulated isoxazolines¹⁵ by using 3-chloro-1,2,4-oxadiazoles as a masked nitrile oxide. Here we report the study of the reaction of 3-chloro-1,2,4-oxadiazoles **1**, with allylamine as an olefin-tethered nucleophile. Selected 3-chloro derivatives **1a–j**, variously substituted at C(5), were obtained from the corresponding 3-amino-1,2,4-oxadiazoles.^{14,16} Compounds **1** were reacted in refluxing methanol in the presence of a 15-fold excess of allylamine giving tetrahydro-isoxazolo[3,4-*d*]pyrimidines **2**, as major products, together with 3-*N*-allylamino-1,2,4-oxadiazoles **3** (Table 1). All reactions were performed for 9 h, except that for compound **1i** (Table 1, entry 9) which reached 100% conversion in 3 h. Complete conversion of the starting material, after 9 h of heating, was not achieved for those derivatives bearing an electron-donating group linked at C(5) of the oxadiazole ring (Table 1, entries 2, 3, 8, 10), while prolonged reflux (> 9 h) caused formation of degradation products.

From a mechanistic point of view, the formation of the observed products **2** and **3** could be ascribed to the presence of two competing reaction pathways both involving an initial attack of allylamine on the oxadiazole ring (Scheme 1).

In route a (Scheme 1), and in analogy to ANRORC reactions reported for 1,2,4-oxadiazoles,^{9–14} the initial nucleophilic attack on the C(5) of the oxadiazole ring causes ring opening into chloro-oxime intermediate **4**. Subsequent elimination of HCl leads to the *in situ* formation of the nitrile oxide intermediate **5** which, through a [3 + 2] heterocyclization, yields tetrahydro-

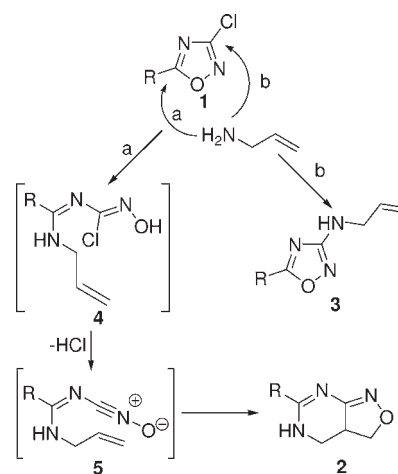
Table 1. Products Distribution for Reaction of Compounds **1** with Allylamine



entry	Reactant	R	2 yield (%) ^a	3 yield (%) ^a
1	1a	Ph	56	35
2	1b (60) ^b	4-MeOPh	21	14
3	1c (58) ^b	4-MePh	16	10
4	1d	4-ClPh	85	14
5	1e	4-NO ₂ Ph	68	18
6	1f	4-CF ₃ Ph	77	19
7	1g	2-furanyl	57	14
8	1h (19) ^b	2-thiophenyl	61	18
9	1i ^c	4-pyridyl	63	33
10	1j (36) ^b	benzyl	40	19

^a Isolated yield. ^b Recovered substrate. ^c Reaction time: 3 h.

Scheme 1. Proposed Mechanism for the Reactions of **1** with Allylamine

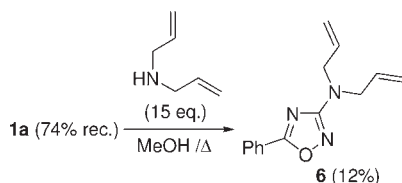


isoxazolo[3,4-*d*]pyrimidines **2**. Conversely, initial attack of the allylamine nitrogen on C(3) (Scheme 1, route b) leads to *N*-allylamino-oxadiazoles **3**, through a classic S_NAr.^{16a} This mechanistic rationale is supported by product distribution data, where compounds bearing electron-withdrawing groups linked at C(5) of the oxadiazole ring gave a higher conversion of starting materials **1** and higher yields of isoxazolo-pyrimidines **2**. In order to consider the effect of the amine structure on the reaction outcome, we have considered the reactivity of representative compound **1a** with *N,N*-diallylamine under the previously reported conditions (9 h, refluxing MeOH) (Scheme 2).

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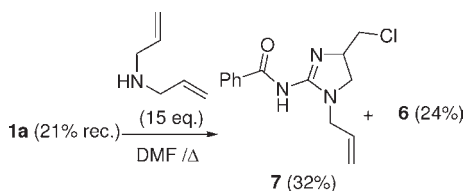
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Scheme 2. Reaction of Compound 1a with Diallylamine



In this case only the S_NAr product **6** was obtained in low yield (12%), thus suggesting that with a hindered nucleophile the ANRORC pathway is suppressed. When the reaction was conducted under more forcing conditions, in refluxing DMF, the yield of compound **6** was doubled and the unexpected imidazoline **7** was obtained as a major product (Scheme 3). From a mechanistic point of view, formation of imidazoline **7** could be explained on the basis of the initial formation of amino-substituted oxadiazole **6** followed by a thermally induced breaking of the O–N bond of the 1,2,4-oxadiazole ring, leading to the nitrene

Scheme 3. Thermal Rearrangement of 1a in the Presence of Diallylamine



intermediate **8** (Scheme 4). Intramolecular aziridination leads to the formation of a bicyclic intermediate **9** (Scheme 4). Intermediate **9** leads to final imidazoline **7** through regioselective aziridine ring opening by action of chloride on the less hindered carbon (Scheme 4, route a), while pyrimidine derivative **10**, obtainable from the competitive attack of chloride, was not observed (Scheme 4, route b) according to previously reported results for aziridine ring opening by a halide ion.¹⁷ Formation of imidazoline **7** was also observed from thermal rearrangement of oxadiazole **6** in refluxing DMF in the presence of an excess of chloride, while thermal treatment of compound **6** in the absence of a chloride ion did not allow isolation of bicyclic intermediate **9**, probably due to decomposition during workup (Scheme 5).

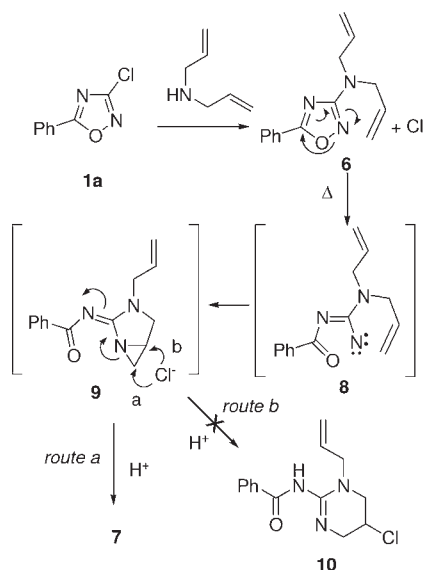
It is interesting to note that the aziridination of N(2) of the 1,2,4-oxadiazole ring was previously observed just as photochemical intermolecular¹⁸ or intramolecular¹⁹ reactivity. Thus, the reported reactivity represents the first

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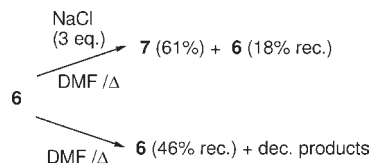
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Scheme 4. Proposed Mechanism for the Formation of Imidazoline 7



Scheme 5. Thermal Rearrangements of 6



entry of thermally induced aziridination of the 1,2,4-oxadiazole ring acting as a nitrene equivalent.

In conclusion, the reactivity of 3-chloro-1,2,4-oxadiazoles with allylamines was investigated, showing two novel thermal rearrangements. Reaction with allylamine yielded tetrahydro-isoxazolo[3,4-*d*]pyrimidines **2**, which, to the best of our knowledge, has not been previously reported in the literature. With respect to previously reported ANRORC rearrangements where the ring-closure step consists of an intramolecular cyclocondensation reaction, a novel tandem ring opening/[3 + 2] dipolar cycloaddition is reported. Moreover, reaction with diallylamine allowed the obtainment of highly substituted imidazoline through an unprecedented thermally induced intramolecular aziridination.

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Supporting Information Available. Synthetic details, characterization data, and ¹H–¹³C NMR spectra of compounds **2**, **3**, **6**, and **7**. This material is available free of charge via the Internet at <http://pubs.acs.org>.